



METHOD OF APPLYING BIOGOMUS AND OBTAINING LIQUID BIOORGANOMINERAL FERTILIZERS FROM EARTHWORM BIOMU

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Abstract: "Gomus" is a Latin word translated as "earth", referring not to the planet itself, but to its upper, fertile layer. It becomes fertile due to the decomposition of organic matter - leaves, grass, animal remains and their metabolic products. Only 5% of the total mass of biohumus is enough for soil fertility. Forming these percentages is not easy. It takes 100 years to create a 10 cm layer. Man-made pollution, unintended use of land slows down the process even more, makes the soil dead.

Keywords: biohumus, mineral fertilizers, potassium, magnesium, magnesium, nitrogen, phosphorus.

Biogomus contains 0.7-1.2% potassium, 0.3-0.5% magnesium, 2-3% magnesium, 0.8-2% nitrogen and a lot of phosphorus. Biogomus for seedlings also contains fulvic and humic acids. Only they are able to process the photon energy of the sun. Acids convert it into chemicals. They block pathogenic bacteria in the soil. It is important for the development and life of plants. Fulvic acid provides cells with the necessary nutrients, prevents swelling, eliminates toxins and viruses.

Many doctors believe that fulvic acid deficiency is one of the causes of any disease in general. It can only be obtained from plants. So, liquid biohumus is not just a fertilizer, but a kind of medicine. For them, the substance is a growth stimulator of the root system. Receiving food, it enters the deep layers of the soil. Moisture can be extracted from them. It is useful during drought. In normal soil, humic acid occurs in water-insoluble form. Plants only absorb solutions. It is possible to dissolve the substance from biohumus.

One of the main problems facing the agriculture of our country is the low coefficient of beneficial effect of mineral fertilizers. It is 60-70% for potassium fertilizers, 20-25% for phosphorus fertilizers in the first year, and 40% in the last 2-3 years.

The next problem is related to soil humus, which is considered the basis of productivity in all soils. Humus plays an important role in the processes of improving the physical properties of the soil and creating a moderate water-weather regime. It acts as an accumulator of soil energy, prevents the leaching of mineral fertilizers from it and thus the widespread pollution of the environment, and also transforms hard-to-dissolve phosphorus compounds into a well-absorbed state. All this ensures the mobility of bound phosphates in the soil, which is from 3 to 6 g of P₂O₅ per 1 m² of cultivated area. This is a large reserve of phosphorus, which is currently not used for crops.

Cultivation of agricultural products is the main source of food production, which is one of the economic sectors. In this place, providing the agro-industrial complex with biological, organic and mineral fertilizer compositions that allow to increase soil fertility, rational use of land and water resources is the main factor of obtaining high-quality and high yield from

agricultural crops. Therefore, one of the priority tasks is to develop a technology for obtaining effective organomineral fertilizers containing humus substances for plant nutrition in agriculture.

Due to the fact that the content of humus in fertile soils is decreasing, special attention is being paid to the production of fertilizers and plant growth stimulants containing humus using organic resources and their use in agriculture. In this regard, the development of technologies for obtaining humic fertilizers and stimulants, which are necessary for increasing soil productivity, is considered one of the urgent tasks.

A certain level of scientific results have been achieved in our republic on the processing of the oxidized form of Angren coal into organomineral fertilizers [1]. In this regard, it is important to establish the production of liquid bioorganomineral fertilizer using earthworm humus as a raw material source. In order to fill the gap in the field of bioorganomineral fertilizers in drip irrigation and hydroponics, a liquid bioorganomineral fertilizer production technology based on earthworm biohumus was created.

The principle technological scheme for the production of liquid bioorganomineral fertilizers is presented in Fig. 1.

5000 kg of water and 500 kg of sieved and ground biohumus are placed in a reactor with an external heater (30-350C) (Figure 1) and a suspension is formed by mixing. 50 kg of nutrient medium (sugar, molasses, etc.) is also added here. The ratio of biohumus to water is 1:10. In order to ensure regular mixing of biohumus with water, a rotating motor is attached to the reactor and a stirrer is attached to its axis of rotation. The rotation speed of the mixer is 180 rpm, the temperature in the reactor water mask is maintained at 30-350C.

The mixture of biohumus with water is periodically stirred in the reactor for 40 minutes. At the same time, air is supplied to the reactor for 120 minutes using a compressor for aeration of aqueous separation from biohumus. After the aeration is completed, the mixed suspension of biohumus is kept in the reactor, where it rests for 2 hours. The suspended bacterial extract on the sediment is pumped from the reactor to the collector (Figure 2).

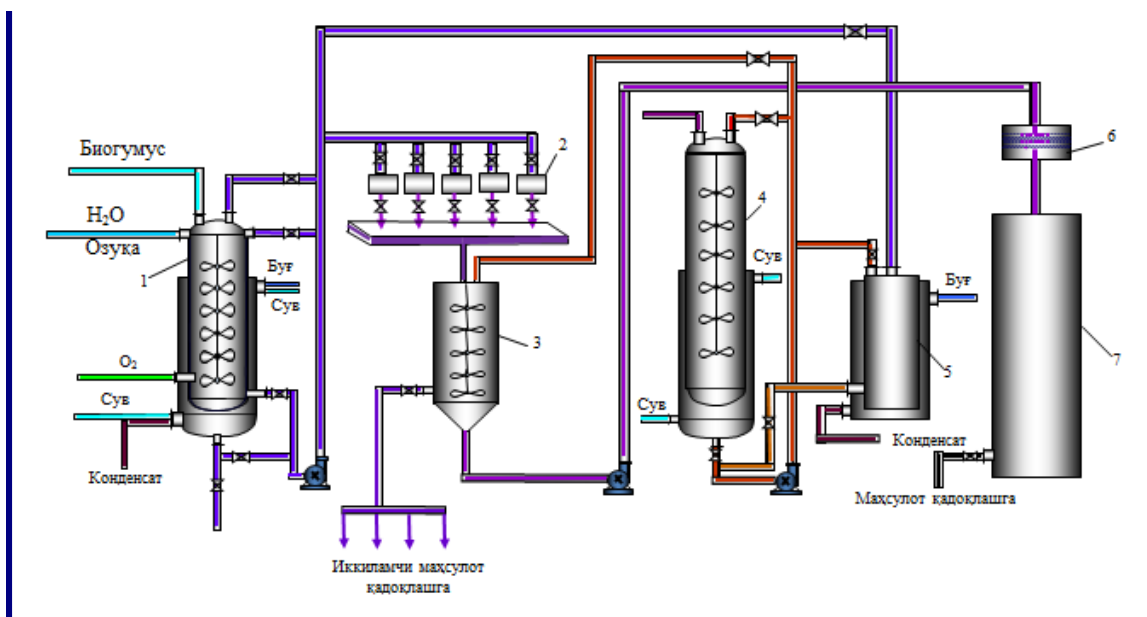


Figure 1. Principle technological scheme of production of liquid bioorganomineral fertilizers from earthworm biohumus:

1st reactor; 2- bacterial secretion collector; 3-liquid bioorganomineral fertilizer preparation mixer; 4-alkaline extract neutralizer; 5-collector-disinfector; 6th filter; 7-finished product warehouse.

5000 kg (1:10 ratio) of boiling water (70-800C) is added to the residual sediment in the reactor (Figure 1). The water temperature in the reactor jacket is also brought up to 70-800C. Then 35 kg of nitric acid (1:0.07 ratio of biohumus) is added to the suspension in the reactor and the process continues for 2 hours with constant stirring. Then the decantation process is carried out - settling and separating the nitric acid solution from the biohumus, then the acidic separation on the sediment is transferred from the reactor to the collector-clarifier (picture 4) using a pump.

5000 kg (1:10 ratio) of boiling water (70-800C) is added to the residual sediment in the reactor (Figure 1). Then, 25 kg of sodium alkali (in a ratio of 1:0.05 to biohumus) is added to the suspension in the reactor, and the extraction process is carried out for 2 hours with constant stirring. In this, the extraction of humic acids from biohumus into an aqueous solution takes place.

Then the decantation process is carried out - settling and separation of the first alkaline solution from the biohumus, then the alkaline separation on the sediment is transferred from the reactor to the collector-clarifier (Figure 5) using a pump.

Boiling water (70-800C) is added to the precipitate in the reactor in a ratio of 1:10 from the heater. During the 2-hour extraction process, stirring is carried out periodically, and 30 kg of potassium alkali (in a ratio of 1:0.06 to biohumus) and 160-200 kg (in a ratio of 1:0.3-0.4) of cotton as a reagent for complete extraction of humic acids are added. ash is added. The extraction process takes 2 hours with constant stirring. In this, complete extraction of humic acids from biohumus into an aqueous solution occurs.

Then, after a 2-hour settling process, a decantation process is carried out - settling and separation of the second alkaline solution from the biohumus.

The mixture of alkaline separations in the collector-cooler (Figure 5) is naturally cooled to a temperature of 20-300C. The cooled alkaline solution is transferred to the accumulator (Figure 4), as a result of which a part of the alkali is neutralized by nitric acid in the accumulator.

The acid-alkaline mixture in the collector (Figure 4) is neutralized by adding small portions of mineral acids until the pH-environment reaches 8-9 with constant stirring, and as a result, a neutralized alkaline solution is formed. Thermal phosphoric acid or extracted phosphoric acid purified from foreign additives (fluorine, sulfate, etc.) is used as mineral acid.

To obtain liquid bioorganomineral fertilizers, the mixture of the bacterial extract from the collector (picture 2) and the neutralized alkaline extract from the collector (picture 4) are mixed in the collector (picture 3) in a ratio of 1:6.

The finished product is passed through a 20-µm filter (Figure 6) and sent to the finished product warehouse (Figure 7) or to the consumer.

After the cycle is completed, the semi-liquid precipitate remaining in the reactor, collectors (images 1, 4 and 5) is transferred to a separate container. It can also be used as a solid organomineral fertilizer [2-30].



Rheological properties of alkaline extract isolated from biohumus and liquid bioorganomineral fertilizer are presented in Table 1.

Table 1

Biogrheological properties of alkaline extract isolated from humus and liquid bioorganomineral fertilizer

Temperature (t), 0S	Alkaline extract		Liquid bioorganomineral fertilizer	
	Density (r), g/cm ³	Viscosity (η), sPz	Density (r), g/cm ³	Viscosity (η), sPz
25	1.0246	1.06	1.0150	1.05
30	1.0223	0.96	1.0135	0.95
40	1.0175	0.83	1.0102	0.81
50	1.0131	0.73	1.0070	0.72
60	1.0090	0.66	1.0037	0.64
70	1.0047	0.61	1.0000	0.58
80	1.0007	0.58	0.9963	0.54

Thus, by adopting this technology in the conditions of industrial production, it is ensured to meet the demand of agro-industrial enterprises for effective liquid bio-organomineral fertilizers with unique consumption properties. These fertilizers are of high quality, can be stored for a long time, can be used for drip irrigation and hydroponics, and show high export potential. The product is high yielding, high value added and agrochemically efficient.

References:

- 1.Беглов Б.М., Намазов Ш.С., Закиров Б.С., Жуманова М.О., Усанбаев Н.Х. Органоминеральные удобрения на основе бурых углей. – Ташкент, 2018 г. – 192 с.
- 2.Шамшидинов, И. Т. (2017). Разработка усовершенствованной технологии производства экстракционной фосфорной кислоты и получения концентрированных фосфорсодержащих удобрений из фосфоритов Каратау и Центральных Кызылкумов. Дисс.... докт. техн. наук, Ташкент.
- 3.Шамшидинов, И. Т. (2014). Технология неорганических веществ и минеральных удобрений: Учебник для профессиональных вузов. ИТ Шамшидинов.
- 4.Мамуров, Б. А., Шамшидинов, И. Т., Усманов, И. И., & Кодирова, Г. К. (2019). Исследование процесса нейтрализации экстракционной фосфорной кислоты мелом. Universum: химия и биология, (2 (56)), 21-26.
- 5.Шамшидинов, И. Т. (1994). Получение удобрений типа двойного суперфосфата из фосфоритов Каратау.
- 6.Gafurov, K., Shamshidinov, I. T., & Arislanov, A. S. (2020). Sulfuric acid processing of high-magnesium phosphates and obtaining NPS-fertilizers based on them. Monograph. Publishing house" Istedodziyo press" Namangan, 26-27.
- 7.Шамшидинов, И. Т. (2017). Исследование процесса переработки фосфоритов Каратау на концентрированные фосфорные удобрения по поточной технологии. Universum: технические науки, (3 (36)), 29-34.

- 8.Кодирова, Г. К., Шамшидинов, И. Т., Тураев, З., & Нажмиддинов, Р. Ю. У. (2020). Исследование процесса получения высококачественных фосфатов аммония из экстрактной фосфатной кислоты на основе фосфоритов Центрального Кызылкума. *Universum: технические науки*, (12-3 (81)), 71-75.
- 9.Нажмиддинов, Р. Ю., Меликүзиева, Г. Қ., Зокиров, М., & Юсупов, И. (2022). Марказий Қизилқум фосфоритларидан таркибида кальций ва магний бўлган концентранган фосфорли оддий ўғитлар олиш. *Ijtimoiy fanlarda innovasiya onlayn ilmiy jurnali*, 2(6), 56-61.
- 10.Shamshidinov, I., Qodirova, G., Mamurov, B., Najmiddinov, R., & Nishonov, A. (2022). Экстракцион фосфат кислотани оҳактош хомашёси билан нейтраллаш асосида кальций ва магний фосфатли ўғитлар олиш. *Science and innovation*, 1(A4), 161-169.
- 11.Najmiddinov, R., Shamshidinov, I., Qodirova, G., Nishonov, A., & Sayfiddinov, O. (2022). Марказий Қизилқум фосфоритлари асосидаги экстракцион фосфат кислотадан юқори сифатли аммоний фосфатлари олиш. *Science and innovation*, 1(A4), 150-160.
- 12.Kodirova, G., Shamshidinov, I., Sultonov, B., Najmiddinov, R., & Mamurov, B. (2021). Investigation of the Process of Purification of Wet-Process Phosphoric Acid and Production of Concentrated Phosphoric Fertilizers Based on it. *Chemical Science International Journal*, 30(1).
- 13.Shamshidinov, I., Qodirova, G. Mamadjanov, Z., Najmiddinov, R. (2021). *International Journal of Advanced Science and Technology*.
- 14.Shamshidinov, I., Qodirova, G. Mamadjanov, Z., Najmiddinov, R. (2021). Экстракция жараёнида фосфат кислотани сульфат ва фтордан тозалаш ҳамда юқори сифатли азот-фосфорли ўғит олишни тадқиқ қилиш.
- 15.Shamshidinov, I., Qodirova, G., Turayev, Z., Mamurov, B. (2020). Study Of The Process Of Heat Treatment Of Limestone To The Process Of Obtaining Calcium-Magnesium-Containing Phosphorous Fertilizers.
- 16.Shamshidinov, I., Qodirova, G., Mamurov, B. (2019). Шўрсув доломитлари асосида кальций ва магний фосфатли ўғитлар олиш. *НамМТИ илмий-техника журнали*.
- 17.Shamshidinov, I., Qodirova, G., Mamurov, B. (2017). Кальций ва магний фосфатли ўғитлар олишда маҳаллий доломит хомашёсидан фойдаланиш.
- 18.28. G'afurov, Q., & Shamshidinov, I. (2010). *Mineral o'g'it ishlab chiqarish nazariyasi va texnologik hisoblari*. T.: Fan va texnologiya, 360.
- 19.G'afurov Q. *Mineral o'g'itlar va tuzlar texnologiyasi: Darslik.*/ Q. G'afurov, I. Shamshidinov. – T.: Fan va texnologiya, 2007. – 360 b.
- 20.Гафуров, К., Шамшидинов, И. Т., & Арисланов, А. С. (2020). Сернокислотная переработка фосфоритов Каратау и сложных удобрений на их основе. Монография. Издательство Lap Lambert Academic Publishing.
- 21.Shamshidinov, I. T., & Mamajanov, Z. N. (2014). Use of low-grade of phosphorites at picking calcium and microelement containing nitrogen-phosphorus fertilizers. *Europaische Fachhochschule*, (3), 117-119.
- 22.Shamshidinov, I. T. Qodirova, G. Najmiddinov, R. Y. (2020). Биогумусдан суюқ биоорганоминерал ўғитлар олиш жараёнини тадқиқ қилиш.
- 23.Гафуров, К., Шамшидинов, И. Т., & Арисланов, А. С. (2020). Сернокислотная переработка высокомагнезиальных фосфатов и получение NPS-удобрений на их основе. Монография.–Наманган: Издательство «Истеъдод зиё пресс».

- 24.Zokirzhon, T., Shamshidinov, I. T., Madamindzanovna, I. O., & Usmanov, I. I. (2019). Researches of the solubility of copper sulfate in orthophosphoric acid at 30 and 80° c. International Journal of Scientific and Technology Research, 8(12), 1870-1872.
- 25.Turaev, Z., Shamshidinov, I. T., Usmanov, I. I., Isakova, O. M., & Sulstonov, B. E. (2019). Thermodynamical Analyse the Formation of Phosphates Copper, Zinc and Cobalt on the Base Double Superphosphate and Sulphates of Copper, Zinc and Cobalt. Chemical Science Internatinal Journal, 28(1), 1-7.
- 26.Shamshidinov, I. T., Gafurov, K. G., & Ikramov, M. M. (2016). Investigation on the phosphoric acid production from low grade phosphorites with high content of magnesium. Journal of Chemical Technology & Metallurgy, 51(2).
- 27.Шамшидинов, И. Т., & Арисланов, А. С. (2022). Влияние магния на процесс экстракции фосфорной кислоты. Central asian journal of theoretical & applied sciences, 3(6), 485-491.
- 28.No, P. 5698 UZ. Method of obtaining extraction phosphoric acid/Gafurov K., Shamshidinov IT, Arislanov A., Mamadaliev A.(UZ)/1998.
- 29.Turgunovich, S. I., & Chorievich, M. K. (2017). Research of process of washing of fluorine from phosphor gypsum. Austrian Journal of Technical and Natural Sciences, (1-2), 107-11.
- 30.Рабинович В.А. Краткий химический справочник/ В.А. Рабинович, З.Я. Хавин. – М.: Химия, 1978. – С.71-79.
- 31.Najmiddinov, R., Shamshidinov, I., Qodirova, G., Nishonov, A., & Sayfiddinov, O. (2022). Марказий Қизилқум фосфоритлари асосидаги экстракцион фосфат кислотадан юқори сифат-ли аммоний фосфатлари олиш. Science and innovation, 1(A4), 150-160.
- 32.Юсупов, И., Зокиров, М., & Сайфиддинов, О. (2022, October). БИОГОМУС ЎҒИТЛА-РИ.
- 33.БИОГОМУСНИНГ ХОССАЛАРИ ВА ҚЎЛЛАНИЛИШИ. In Международная конференция академических наук (Vol. 1, No. 29, pp. 17-24).
- 34.Сайфиддинов, О., Ғойипов, А., & Рахмонов, Д. (2022). Композицион фенолформальдегид смолаларини термик хоссаларини ўрганиш. Zamonaviy dunyoda innovatsion tadqiqotlar: Nazariya va amaliyot, 1(23), 99-102.
- 35.Турсунбоев, Х., & Сайфиддинов, О. (2022). ИСПОЛЬЗОВАНИЕ ИННОВАЦИОН-НЫХ ТЕХНОЛОГИЙ В ОБУЧЕНИИ СТУДЕНТОВ НЕХИМИЧЕСКИХ НАПРАВЛЕНИЙ. Zamonaviy dunyoda innovatsion tadqiqotlar: Nazariya va amaliyot, 1(28), 434-438.
- 36.Юлдашева, Н., Сиддиқова, Ў., Тўхтасунова, М., & Сайфиддинов, О. (2023). ГАЗОВЫЕ ДАТЧИКИ ДЛЯ ПРИРОДНОГО И ГОРЯЧЕГО ГАЗА ХИМИЧЕСКИЕ ДАТЧИКИ ЗОЛ-ГЕЛЬ СИНТЕЗ НАНОКОМПОЗИТНЫХ ПЛЕНОК. Естественные науки в современном мире: теоретические и практические исследования, 2(4), 18-20.
- 37.Eminov, A., Jumanov, Y. U., Umarov, F., & Sayfiddinov, O. (2022). PROSPECTS FOR THE USE OF KAOLINS OF UZBEKISTAN. Science and Innovation, 1(6), 367-373.
- 38.Сайфиддинов, О. (2023). СОДЕРЖАЩИЕ ЦЕНТРАЛЬНЫЕ КРАСНЫЕ ФОСФОРИ-ТЫ КОНЦЕНТРИРОВАННЫЙ ФОСФОР С КАЛЬЦИЕМ И МАГНИЕМ ПОЛУЧЕНИЕ ПРОСТЫХ УДОБРЕНИЙ. Евразийский журнал академических исследований, 3(7), 213-215.
- 39.Najmiddinov, R., Shamshidinov, I., Kodirova, G., & Sayfiddinov, O. (2023). MARKAZIY QIZILQUM FOSFORITLARI ASOSIDA FOSFOR KISLOTANI KALSIY KARBONATI BILAN EKSTRAKTSIYA VA BITTA FOSFORLI O'GITDAN OLISH JARAYONDA TOZALASH. Tsentralnoaziatskiy jurnal akademicheskix issledovaniy , 1 (2), 47-57.
- 40.Eminov, A. M., Sh, U. F., & Sayfiddinov, O. O. (2023). Mahalliy kaolinlar asosida qurilish keramikasi materiallari olish istiqbollari. Строительство и образование, 1(3), 22-27.

41. Сайфиддинов, О., & Усканбеков, О. (2022). НАНО-ЗАРРАЧАЛАРНИНГ ХОССАЛА-РИИ ЎРГАНИШ. Инновационные исследования в современном мире: теория и практика, 1(28), 18-22.

